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2 IMPORTANT NOTE: To compare the strength of acids, we examine their conjugate bases. A base is stronger when it is able to share its electron pair. Because they influence electron density, the following structural effects influence basicity. When a factor takes electron density from the atom that wants to share

BrÃ¸nsted-Lowry Lewis Acid Base - UCLA Chemistry and

Reference: Hardinger, Steven. Chemistry 14C: Structure of Organic Molecules : Course Thinkbook : Concept Focus Questions, OWLS Problems, Practice Problems.

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